Directions: Choose the letter of the correct answer.

- 1. $\frac{31}{15}P$ and $\frac{32}{16}S$ are related as
 - a. isobars
 - b. isotones
 - c. isotopes
 - d. none of the above
- 2. Which of the following is the correct identity of $\frac{A}{Z}X$ in the nuclear reaction below?

$$^{25}_{12}Mg + {^4}_{2}He \rightarrow {^A}_{7}X + {^1}_{1}H$$

- b. $\frac{^{28}}{_{13}}Si$ c. $\frac{^{28}}{_{14}}Al$ d. $\frac{^{28}}{_{14}}Si$
- 3. The reaction below is an example of:

$$^{212}_{84}Po \rightarrow ^{208}_{82}Pb + ^{A}_{Z}X$$

a. beta decay



Nuclear Chemistry

Practice Questions

- b. alpha decay
- c. positron decay
- d. electron capture
- e. gamma radiation
- 4. How many half-lives are necessary to reduce 131 I ($t_{1/2}$ = 8 days) to a quarter of its original amount?
 - a. 2
 - b. 8
 - c. 16
 - d. 32
- 5. Which of the following terms will give the correct rate constant for the spontaneous decomposition of 26 Na ($t_{\frac{1}{2}}$ = 1.1 s)?
 - a. $\frac{\ln 2}{26 s}$
 - b. $\frac{\ln 2}{1.1 \, s}$
 - C. $\frac{26 \text{ s}}{\ln 2}$
 - d. $\frac{1.1 \text{ s}}{\ln 2}$